

# Formation energy of charge states of nitrogen and oxygen vacancies in anatase TiO<sub>2</sub>: An ab initio study

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Electronic and structural properties of several charge states of interstitial (N<sub>i</sub>) and substitutional (N<sub>s</sub>) nitrogen and oxygen vacancies (V<sub>O</sub>) into anatase TiO<sub>2</sub> were studied through density functional theory calculations.

The formation energies indicate that different charged states happen in the range of allowed electronic chemical potential,  $\mu_e$ . For N-doping, neutral and positively charged states occur at  $\mu_e$  near the valence band top, namely N<sub>s</sub><sup>0</sup>, N<sub>s</sub><sup>+</sup> and N<sub>i</sub><sup>2+</sup>, while negatively charged states occur at  $\mu_e$  near the conduction band bottom, N<sub>s</sub><sup>-</sup> and N<sub>i</sub><sup>-</sup>.

N<sub>i</sub> is energetically more stable than N<sub>s</sub> and always has a negative formation energy. The presence of oxygen vacancies would facilitate the formation of N<sub>s</sub><sup>-</sup> because the defect levels associated to V<sub>O</sub> can pin the Fermi level close to the bottom of the conduction band. The neutral and charged states of V<sub>O</sub> always show positive formation energies. Vacancies with charge state 2+ have the highest stability, as has been reported in the literature.