Colloidal Stability has a crucial role for nanomaterials toxicity testing *in-vitro*: nZVI-*algae* colloidal system as case study

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Abstract

Aggregation is one among those parameters focusing the attention in Nanotoxicology due to its methodological implications [1, 2]. Aggregation is a physical symptom of a more general physicochemical condition of colloidal particles, that is, colloidal stability. A destabilized colloidal system may tend to reduce its net surface energy by self-aggregation, but also by hetero-aggregation which may involve biological interfaces. In this regards, colloidal stability may have an important role as driver of ENM bioactivity. In the present study, a physical speciation phenomenon of nZVI nanoparticles, we called colloidal singularity [3], was found when generating a dose gradient of zero-valent iron nanoparticles (nZVI) in algal culture medium. The colloidal singularity consisted of an exceptionally stable ENM suspension with particles in their primary size (4 - 12 nm) occurring within a tight dose range (0.1-0.5 mg/L). Outside this range, nZVI suspensions aggregated. Interestingly, nZVI exhibited toxicity to the algal model organism, except in the 0.1-0.5 mg/L dose range. Stability analyses, TEM images and FTIR revealed that nZVI toxicity was mediated by nZVI-alga interaction, and that the increased colloidal stability of nZVI suspensions in the 0.1-0.5 mg/L dose range prevented nZVI-algae interaction and subsequent toxicity. Furthermore, in-situ destabilization of this particular suspension using a classical flocculating agent Al₂(SO₄)₃ resulted in toxicity. These observations demonstrate that colloidal stability has a major role in nZVI toxicity and that may be carefully taken in to account when performing safety assessment of nanomaterials beyond the initial considerations for stock sample preparation.

References

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