

The effect of hole-electron for decolorization of dyes by using nanomagnetic compound

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The Graphene oxide (GO) has a plethora of composites which are constituted by various nanomaterials like Fe₃O₄ as a strong nanomagnetic compounds and could increase the properties of both materials correlatively which have been used for many ends such as drug delivery systems, electronic devices, and wastewater treatments these days. The photocatalytic activities of Fe₃O₄-GO nanocomposites were observed under UV and visible light illumination. The removal yields of Reactive Red 198 (RR198) at various conditions are showed approximately 100% of RR198 was degraded during 60 min in the presence of photocatalysts due to UV-induced with assist of hydrogen peroxide to photolysis of dye and a negligible decrease in dye concentration was obtained afterwards which are depicted in fig. 1 and fig. 2. Near 100% of the dye was decolorized under UV exposure in different samples which were arranged to examine in order to explore modify condition for decolorization. According to these samples, The RR198 was examined by different conditions which were observed: RR198/UV, Fe₃O₄-GO/UV, H₂O₂/UV, Fe₃O₄-GO/H₂O₂-UV, and Fe₃O₄-GO/H₂O₂-visible light, Fe₃O₄-GO/visible light, and H₂O₂-visible light. The Fe₃O₄-GO behave as a strong photocatalyst could create hole-electron on its surface and produce the free radical of hydroxyl on its surface and assist to augment and intensify severely the effect of decolorization in aqueous solutions. Decolorization is not only arranged with a low dosage of H₂O₂ in volume but also it is equipped by low power of UV irradiation in only 25W. The Fe₃O₄-GO could produce •OH and •O₂⁻ which are assisted indirectly to disappear the RR198 easily. Fe₃O₄-GO manners such a strong photocatalyst to remove RR198 under UV and in absence of UV irradiation which are discovered by these examinations and methods.

Keywords: Fe₃O₄-GO, decolorization, nanomagnetic compound.

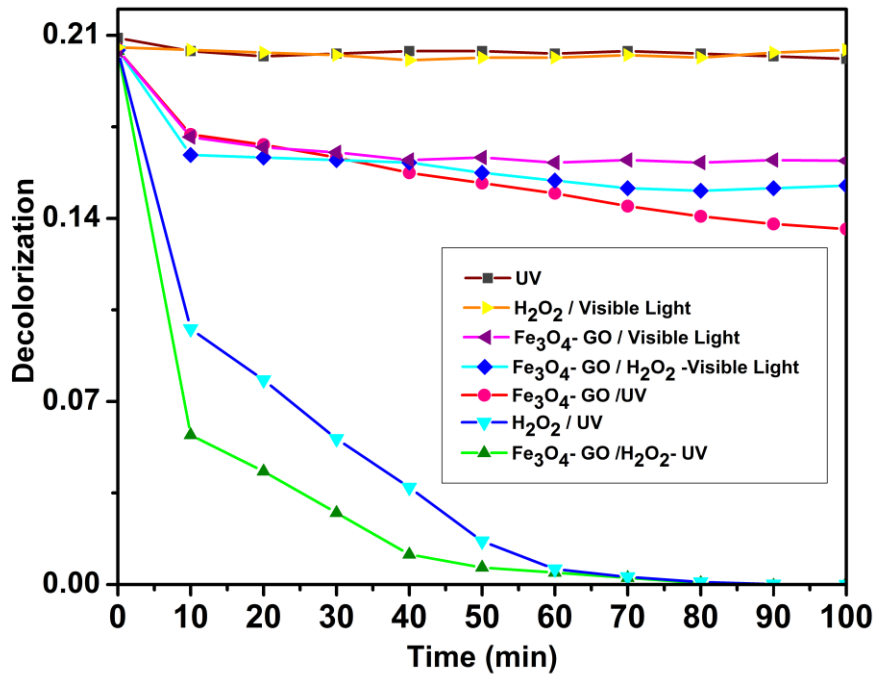


Fig. 1. The UV irradiation of RR198 with various conditions during 100 min.

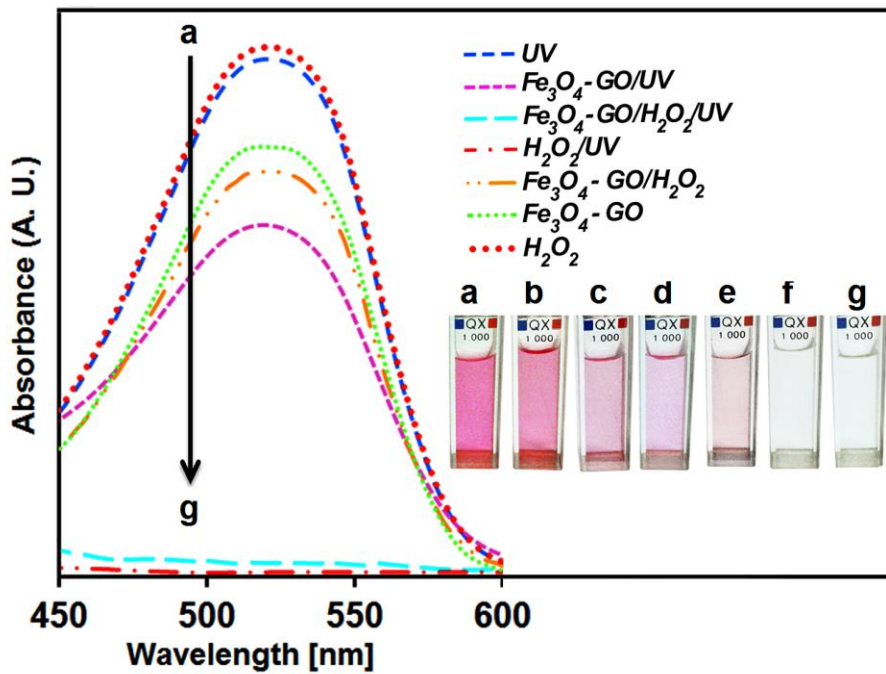


Fig. 2. The UV-Vis spectroscopy and Decolorization of RR198 in different conditions.