

## Non-covalent Interactions to Graphene: Theory and Experiment

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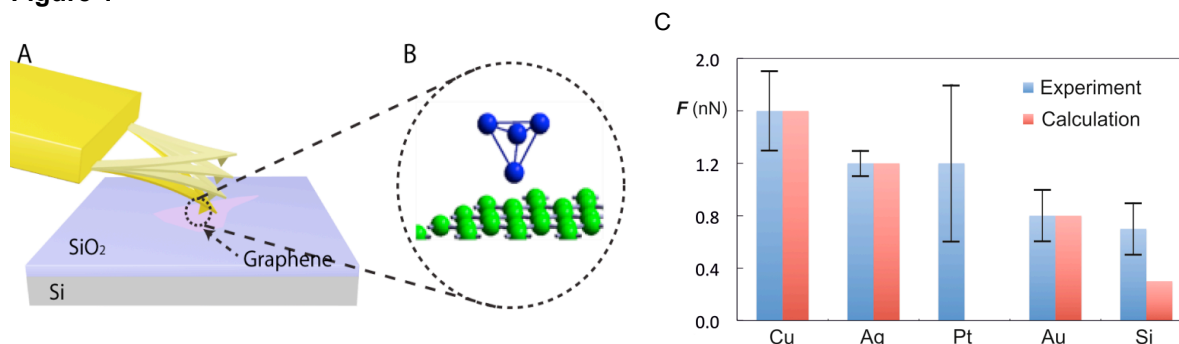
### Abstract

Graphene is a two-dimensional  $\pi$ -conjugated material having extraordinary physical properties, which makes it a perspective material in catalysis, energy storage, nano(opto)electronics and sensor applications.[1,2] The application potential of graphene can be enormously enhanced by its covalent and non-covalent functionalization.[3] An exact quantification of interaction between graphene and guest molecules as well as thorough understanding of the nature of interaction between graphene and guest molecules have not been yet achieved. We analysed nature of interaction between Ag, Au, Pd, Pt metal atoms and clusters to benzene, coronene and graphene, by quantum chemical calculations [4,5] The silver atom is bound weakly by London dispersion forces, while interaction of palladium and platinum is significantly stronger and involves some covalent character. Involvement of relativistic effects is required for a reasonable description of interactions involving Au and Pt. We measured the interaction force between metallized AFM tips (Cu, Ag, Au, Pt and naturally Si were considered, Figure 1) and graphene and correlated the results with theoretical calculations carried out by density functional theory (DFT). The theoretically calculated interaction forces agree with the experimental data, only when non-local electron correlation and exact Hartree-Fock electron exchange is explicitly treated.[6] The AFM tip covered by copper displayed the highest affinity to graphene among the metals considered. We also quantified the adsorption enthalpies between graphene and several organic molecules and compared the experimental results with theoretical calculations carried out by DFT and spin-component scaled MP2 calculations.

### References

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- [2] Novoselov KS et al. *Nature* **490**, (2012) 192.
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Figure 1



A: The interaction force between graphene and metallized AFM tip was measured by dynamic AFM at ambient conditions. B The model used for theoretical calculations involves graphene (green) and metal tetrahedron (blue). C: Comparison between measured and calculated forces. Adopted from Ref. [6].