Spontaneous electric polarization of a two-dimensional hexagonal lattice: Do we correctly understand the ground state of graphene?

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The static dielectric constant of bulk (three-dimensional) insulators is related to the atomic polarizability α of constituting atoms by the known Clausius-Mossotti formula. We derive a similar expression for a two-dimensional crystal with the hexagonal lattice. The result shows that due to the local field effects the effective *macroscopic* polarizability λ of such a crystal becomes negative under the condition

$$\beta = \alpha/2a^3 > \beta_{crit} = 0.02925$$

where *a* is the lattice constant. For the carbon atom the calculated value of α_{C} is [1]

$$\alpha_{\rm C} = 1.54 \,{\rm \AA}^3$$

while the lattice constant of graphene is 2.46 Å. The β -parameter for graphene

 $\beta_{graphene} \approx 0.05 > \beta_{crit}$

is thus larger than the critical value of the sign inversion of λ .

A two-dimensional crystal with the negative macroscopic polarizability is thermodynamically unstable which means that intrinsic graphene should have a spontaneous electric polarization, i.e to be in the ferroelectric ground state.

The spectrum of electrons in graphene is usually described within the tight-binding approximation [2] which was originally used for three-dimensional graphite [3]. However in three dimensions the sign inversion of the macroscopic polarization is not the case as follows from the Clausius-Mossotti formula. The two-dimensional crystals thus substantially differ from three-dimensional ones in this respect and the essentially single-particle tight-binding approximation should be applied to them with care. Since it neglects electron-electron interaction between π -electrons the predicted here effect cannot be obtained within the tight-binding approach even if the interaction was taken into account as perturbation. Theoretical results obtained for graphene so far may thus need to be reconsidered.

Since the atomic polarizability of B and N atoms are of the same order as that of carbon [4], the same conclusions are valid for boron-nitride monolayer crystals.

References

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