

# Designer sub-nm ion channels

**K. Gopinadhan<sup>1</sup>**

A. Esfandiar<sup>1</sup>, F. C. Wang<sup>2</sup>, B. Radha<sup>1</sup>, and A. K. Geim<sup>1</sup>

<sup>1</sup>School of Physics and Astronomy, University of Manchester M13 9PL, UK

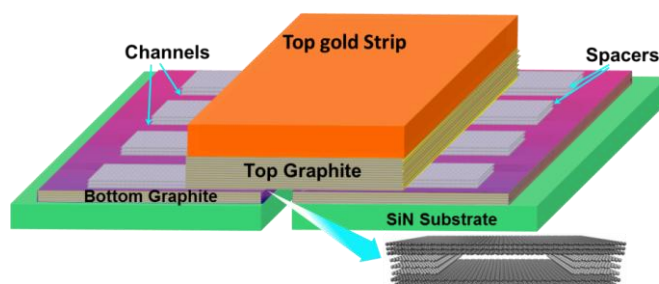
<sup>2</sup>Chinese Academy of Sciences Key Laboratory of Mechanical Behavior and Design of Materials, Department of Modern Mechanics, University of Science and Technology of China, Hefei, Anhui 230027, China

[kalon.gopinadhan@manchester.ac.uk](mailto:kalon.gopinadhan@manchester.ac.uk)

Due to severe problems of accessible clean water, the world is looking at alternate technologies/mechanisms by which efficient separation of ions and molecules at the smallest length scale can be made possible. The importance of membranes with good filtration capabilities in the sub-nm scale is ever increasing. In this regard, carbon nanotubes, graphene nanopores, and graphene oxide have been tried; however, the large distribution of sizes and limited possibility for large scale integration makes these systems not very popular for technologies. Recently, we have reported a novel approach of utilizing the atomic flatness of van der Waals layered materials for the fabrication of sub-nm channels, beating the surface roughness limit traditionally encountered in lithographically fabricated systems [1]. This technique utilizes a combination of van der Waals assembly of layered materials and lithography to create fluidic channels of sizes 'at will' and 'choice'. We systematically studied the ion transport through these sub-nm channels and the schematic of the device is shown in figure 1. We present some of the interesting effects at sub-nm scale.

## References

- [1] Radha, B. et al. *Nature* **538**, 222-225 (2016).



**Figure 1:** Schematic of the nanochannel device utilized for ion transport study. The width of the channel is 130 nm and the height can be varied from 1 atomic layer to several layers.