

All-carbon Solid State Supercapacitors Based on Graphene

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Abstract

Graphene-based supercapacitors attracted a lot of interest due to the advantages of graphene over activated carbon and carbon nanotubes (CNTs) in energy storage. This energy storage technology is poised for an explosive growth, due to the high power density and to the flexibility of the material, which can be adapted to a number of applications, from mobile phones to electric cars. Conventional supercapacitors contain metal foils as current collectors, which suffer from corrosion by the acid or alkali electrolyte. We produced all-carbon solid-state supercapacitors using graphene electrodes and double-wall CNT films as current collectors. Though CNTs are relatively inferior compared to graphene in achieving high capacitance, their high electrical conductivity and 1D structure can be useful to replace the metal when combined with graphene for making supercapacitors. The graphene was produced by a simple ultrasound assisted electrochemical exfoliation method. A very low resistance ($5 \Omega/\text{sq}$) DWCNT film was coated onto plastic substrates. A thin layer of gelled electrolyte is sandwiched by two electrode films, therefore the whole device is flexible. High capacitance, around $10 \text{ mF}/\text{cm}^2$, is achieved with an electrode film less than $10 \mu\text{m}$ in thickness. The energy density of our devices is in the order of $10^{-3} \text{ Wh}/\text{cm}^3$, comparable to that of thin film Li battery. The power density is around $10 \text{ W}/\text{cm}^3$ [1] which is about 10 times higher than other works such as those based on laser-scribed graphene [2].

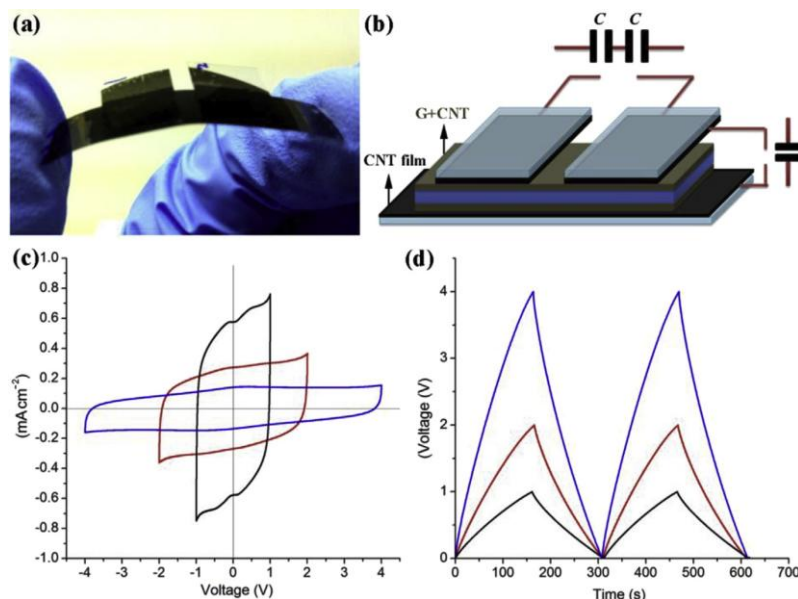


Fig 1 (a), (b) in-plane assembly of two supercapacitors. (c) CV curves measured at 60 mVs^{-1} for a single, two and four devices in series. (d) CD curves measured at 50 mAcm^{-2} for a single device, two, and four devices in series.

References

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- [2] M. F. El-Kady and R. B. Kaner, *Nature communications*, vol. 4, p. 1475, 2013.